

## Bookmark File PDF Hydroxycinnamic Acid Antioxidants An Electrochemical Overview

# Hydroxycinnamic Acid Antioxidants An Electrochemical Overview

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Dr. Jeffry Gerber at Ketofest 2017 - Cholesterol OMG!~~The Key to Stimulating Soil Biology Capturing Residue to Build Soil Organic Matter~~ Holy Grail of Crop Health: Plant Secondary Metabolites - by Jerry Brunetti ~~Hydroxycinnamic Acid Antioxidants An Electrochemical~~ Hydroxycinnamic acids have gained an increasing interest in health because they are known to be potent antioxidants. These compounds have been described as chain-breaking antioxidants acting through radical scavenging activity, that is related to their hydrogen or electron donating capacity and to the ability to delocalize/stabilize the resulting phenoxyl radical within their structure.

## ~~Hydroxycinnamic acid antioxidants: an electrochemical overview~~

Hydroxycinnamic Acid Antioxidants: An Electrochemical Overview 1. Introduction. In the last decade, dietary polyphenols, which are the most abundant antioxidants present in a human... 2. Hydroxycinnamic Acids: Classification and Occurrence. Hydroxycinnamic acids (HCAs) possess a simple chemical ...

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## Hydroxycinnamic Acid Antioxidants: An Electrochemical Overview

Hydroxycinnamic acids (such as ferulic, caffeic, sinapic, and p-coumaric acids) are a group of compounds highly abundant in food that may account for about one-third of the phenolic compounds in...

(PDF) Hydroxycinnamic Acid Antioxidants: An ...

3. Hydroxycinnamic Acids: An Antioxidant Outlook Antioxidants, used to prevent or inhibit the natural phenol-oxidation, have a broad application in diverse fields as they have a huge importance either as industrial additives or health agents. In this context, HCAs have been ascribed to act as powerful antioxidant compounds possessing diverse

## Hydroxycinnamic Acid Antioxidants: An Electrochemical Overview

Hydroxycinnamic acids (such as ferulic, caffeic, sinapic, and p-coumaric acids) are a group of compounds highly abundant in food that may account for about one-third of the phenolic compounds in our diet. Hydroxycinnamic acids have gained an increasing interest in health because they are known to be potent antioxidants.

Hydroxycinnamic Acid Antioxidants: An Electrochemical ...

Hydroxycinnamic Acid Antioxidants: An Electrochemical Overview Table 4 Redox potentials and antioxidant activity of hydroxycinnamic acids, ester and amide derivatives. \*The results

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of DPPH assays are usually expressed as TEAC (trolox equivalent antioxidant capacity) or IC<sub>50</sub> (concentration which is required to scavenge 50% of DPPH free radicals).

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## ~~Hydroxycinnamic Acid Antioxidants: An Electrochemical Overview~~

Considering that hydroxycinnamic acids are antioxidants compounds by excellence, electrochemical techniques can be powerful tools for the study of reaction mechanisms involving electron transfer and aord comple- mentary information. emain structural feature responsible for the antioxidant and free radical-scavenging activity of hydroxycinnamic acid derivatives is the number and location of hydroxyl groups present in the molecule.

## ~~Review Article Hydroxycinnamic Acid Antioxidants: An ...~~

Hydroxycinnamic acid antioxidants: An electrochemical overview

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## ~~Hydroxycinnamic Acid Antioxidants An Electrochemical Overview~~

Measurements of the electrochemical behavior can be excellent guide with a lot of useful information about antioxidant activity of hydroxycinnamic acids [25-31]. During the analysis of results rated the relationship between the measured potential and antioxidant properties. Electrochemical studies in

## ~~Determination of Antioxidant Activity of Caffeic Acid and ...~~

Abstract Hydroxycinnamic acids (HCs) (coumaric acid, ferulic acid, sinapic acid, caffeic acid, chlorogenic acid, rosmarinic acid) are phenolic compounds found in fruits, vegetables, and beverages (coffee, tea, wine). HCs are of particular interest because of their biological properties and potential applications.

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## ~~Antioxidant Properties of Hydroxycinnamic Acid Derivatives ...~~

Hydroxycinnamic acids (HCAs), namely rosmarinic acid, para-coumaric acid, caffeic acid, ferulic acid and synapic acid, have a phenylpropanoid structure, which consists of an aromatic ring bearing different substituents (most often hydroxyl or methoxy groups) and a propane.

## ~~Electrochemical Methods and (Bio) Sensors for Rosmarinic ...~~

One of the most important HCA derivatives is chlorogenic acid (CGA) which has been reported as an efficient antioxidant agent [56, 57]. Chlorogenic acids (CGAs) are esters of HCAs and quinic acid. The most common CGA is formed by esterification of caffeic acid to quinic acid (Fig. 2).

## ~~Antioxidant Properties of Hydroxycinnamic Acids: A Review ...~~

These hydroxycinnamic acids have an important role on the beverage taste and quality of coffee beans and exhibit prominent antioxidant activity (Vignoli et al., 2014). These polyphenols have called attention due to their ability to scavenge radicals, thus restoring oxidative balance in physiological systems (Parras, Martínez-Tomé, & Jiménez, 2007).

## ~~Electrochemical behavior and determination of major ...~~

Hydroxycinnamic acids are the most widely distributed phenolic acids in plants. Broadly speaking, they can be defined as compounds derived from cinnamic acid. They are present

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at high concentrations in many food products, including fruits, vegetables, tea, cocoa, and wine.

~~Biosynthesis, Natural Sources, Dietary Intake ...~~

~~Electrochemical Behavior and Antioxidant and Prooxidant Activity of Natural Phenolics ...~~

~~Overlapped cyclic voltammograms of 1 mM p-coumaric acid, as a typical mono-hydroxycinnamic acid, obtained at scan rates of 25, 50, 100, 200, 400 and 500 mVs<sup>-1</sup>. Insert A shows linear plot of anodic peak currents versus square root of scan rate.~~

~~Electrochemical Behavior and Antioxidant and Prooxidant ...~~

~~Hydroxycinnamic acid amide derivatives, phenolic compounds and antioxidant activities of extracts of pollen samples from Southeast Brazil. Journal of Agriculture and Food Chemistry, 59 (10), 5516-5522.~~

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