

## Chemistry Unit 6 Test Answer Key

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### Unit 6 - Test(Chemical Compound) - ProProfs Quiz

The lines/curves indicate the maximun amount of solute which can be dissolved in 100g of water at each temperature. We call this a saturated solution.. Any amount above the curve is said to be supersaturated; there is more solute than the solvent can hold, therefore the excess precipitates out/settles to the bottom.. Any amount below the line is called unsaturated, meaning more solute can be ...

### Unit 6 | chemistry

UNIT 6: RADIOACTIVITY AND NUCLEAR CHEMISTRY 5. A skeleton of a mammoth was analysed and found to contain 21.2% of the carbon-14 found in living bone. The half-life of carbon-14 is 5730 years. What is the most likely year of the mammoth's death? Use  $\log(\text{Ni N}) = t \log 2$   $t = \frac{1}{2} \log \frac{\text{Ni N}}{\text{Ni N}}$   $t = \frac{1}{2} \log \frac{100}{21.2}$   $t = \frac{1}{2} \log 4.717$   $t = \frac{1}{2} \times 0.674$   $t = 0.337$  years; hence find t This year – t = year of death

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UNIT 6 – REDOX REACTIONS 6 • The oxidation number of an atom is the charge that would exist on an individual atom if the bonding were completely ionic • In simple ions, the oxidation number of the atom is the charge on the ion: - Na+, K+, H+ all have an oxidation number of +1 - Mg2+, Ca2+, Pb2+ all have an oxidation number of +2 - Cl-, Br-, I-all have an oxidation number of -1

### UNIT 6 REDOX REACTIONS - A-Level Chemistry

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### GCSE Chemistry (Single Science) - BBC Bitesize

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